## PRELIMINARY NOTE Iodine-catalyzed Rearrangement of Polyfluorinated 2-(Dichloroamino)pyridines

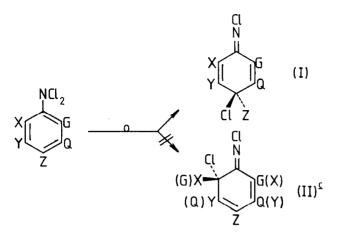
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Numerous examples of iodine-catalyzed or 'spontaneous' rearrangement of fluorine-containing <u>NN</u>-dichloroanilines to <u>N</u>,4-dichlorocyclohexadienylideneamines have been studied [1-4], and in each case formation of only one [<u>viz</u>. the 2,5-isomer (I)] of the two possible chloriminocyclohexadienes has been observed (see Scheme 1). <u>N</u>-Chlorination of 2-aminoheptafluoronaphthalene [6] appeared to provide only the corresponding <u>N</u>-chloro-imine containing a 2,4-dienic moiety [as in (II)], which is not unexpected since 'para-migration' of an <u>N</u>-chloro substituent would have disrupted the aromatic sextet in ring A (see Scheme 2). In the pyridine series, attempts to force 4-(dichloroamino)tetrafluoropyridine [7] or 3-chloro-4-(dichloroamino)trifluoropyridine [3] to isomerize led to the formation of the corresponding azocompounds.

As part of an investigation of <u>skeletal</u> rearrangements which occur when <u>N</u>,4-dichlorocyclohexa-2,5-dienylideneamines are thermolyzed [8], work aimed at the synthesis of azaanalogues of the 2,5-dienes (I) was undertaken. This has provided the first examples of iodine-catalyzed rearrangement of monocyclic dichloroaminoarenes to chloriminocyclohexa-2,4-dienes (see Scheme 3). A single attempt to isolate

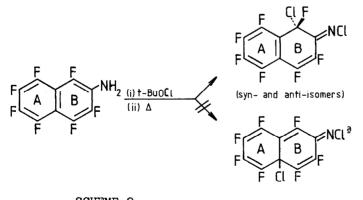
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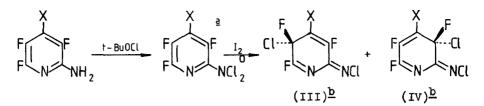
SCHEME 1 [5]

<sup>a</sup>Iodine-catalyzed.

<u>b</u>'Spontaneous' rearrangement occurs during attempted isolation. <u>C</u>None detected by <sup>19</sup>F n.m.r.



<u>SCHEME 2</u> <u>a</u>None detected by <sup>19</sup>F n.m.r.



X = H, (III) : (IV) = 94 :  $6^{\underline{C}}$ X = Cl, (III) ; (IV) = 90 :  $10^{\underline{C}}$ X = CF(CF<sub>3</sub>)<sub>2</sub>, (III) : (IV) = 60 :  $40^{\underline{C}}$ 

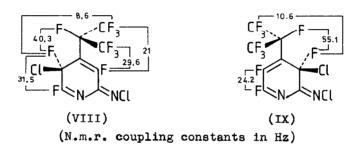
## SCHEME 3

<sup>A</sup>Yields: X = H, 96%; X = Cl, 82%; X = CF(CF<sub>3</sub>)<sub>2</sub>, 77% <sup>D</sup>Combined yields of (III) + (IV): X = H, 86%; X= Cl, 87%; X = CF(CF<sub>3</sub>)<sub>2</sub>, 60% <sup>C</sup>Ratios determined by <sup>19</sup>F n.m.r.

2-(dichloroamino)-3,5,6-trifluoro-4-methylpyridine following treatment of the corresponding  $NH_2$ -compound with t-butyl hypochlorite failed, removal of solvent (CCl<sub>4</sub>) at the end of the reaction providing a mixture of the 2,5-diene (III; X = Me) and, through attack on that product by adventitious moisture, the amide (V). Treatment of the mixtures (III + IV) where X = H or Cl with water rapidly gave amides (VI) and (VII), the corresponding products derived from nucleophilic displacement of the imidoyl fluorine from the minor isomer (IV) in each case not being detected by <sup>19</sup>F n.m.r. spectroscopy.

(V) X = Me
(VI) X = H
(VII) X = Cl

The structures of the 6-chlorimino-1-azacyclohexadienes (III) and (IV)  $[X = H, Cl, \text{ or } CF(CF_3)_2]$  were determined by <sup>19</sup>F n.m.r. spectroscopic examination of their mixtures<sup>\*</sup> [only in the case of the 4-chloro system was it possible to isolate (by g.l.c.) a pure sample of the major isomer (III; X = Cl)]. Interestingly, the 4-heptafluoroisopropyl-compounds (III) and (IV)  $[X = CF(CF_3)_2]$  are judged to be 'locked' conformationally [see (VIII) and (IX)] with non-equivalent CF<sub>3</sub> groups in each isomer and substantial 'through space' coupling of the tertiary fluorine to the adjacent olefinic fluorine; this can be ascribed to the buttressing effect of the chlorines of the adjacent CFC1 groups.



Further work is in progress to determine the factors affecting the ratio of 'ortho' (--chloriminocyclohexa-2,4dienes) to 'para' (--chloriminocyclohexa-2,5-dienes) migration of <u>N</u>-chlorine substituents during iodine-catalyzed rearrangements of <u>NN</u>-dichloroamino-arenes and -hetarenes.

## REFERENCES

- 1 R.E. Banks and T.J. Noakes, J.C.S. Perkin Trans. I, (1976) 143.
- 2 R.E. Banks, M.G. Barlow, and T.J. Noakes, J. Fluorine Chem., 10 (1977) 45.

<sup>\*</sup>Correct elemental analyses were obtained for the mixtures and for all the other new compounds encountered in this work.

- 3 R.E. Banks, M.G. Barlow, T.J. Noakes, and M.M. Saleh, J.C.S. Perkin Trans. I, (1977) 1746.
- 4 R.E. Banks, M.G. Barlow, J.C. Hornby, and T.J. Noakes, J. Fluorine Chem., <u>13</u> (1979) 179.
- 5 Russian workers [Yu.G. Shermolovich, O.M. Polumbrik, L.N. Markovskii, E.P. Saenko, G.G. Furin, and G.G. Yakobson, J. Org. Chem. U.S.S.R., <u>13</u> (1977) 2404] have checked our [1] observations on the iodine-catalyzed rearrangement of <u>NN</u>-dichloroperfluoroaniline and, independently (<u>cf.</u> [3]), studied the analogous rearrangement of <u>NN</u>-dichloroperfluoro-<u>p</u>-toluidine; they too failed to detect the formation of the corresponding 2,4-dienes (II; X = Y = Z = Q = G = F; X = Y = Q = G = F,  $Z = CF_3$ ).
- 6 R.E. Banks, M.G. Barlow, and M.M. Saleh, J. Fluorine Chem., <u>10</u> (1977) 81.
- 7 R.E. Banks and T.J. Noakes, J.C.S. Perkin Trans. I, (1975) 1419.
- 8 B. Al-Saleh, R.E. Banks, M.G. Barlow, and J.C. Hornby, J. Fluorine Chem., <u>12</u> (1978) 341.